



1. Please solve the following equations:

- (a)  $x^2y'' - 2xy' + 2y = 0$  (8%)
- (b)  $y'' - 4y' + 4y = 0, \quad y(0) = 2, \quad y'(0) = 1,$  (8%)
- (c)  $xe^y y' + (2x + e^y) = 0$  (8%)
- (d)  $y''' - 2y'' - y' + 2y = 2x^2 - 6x + 4, \quad y(0) = 5, \quad y'(0) = -5, \quad y''(0) = 1,$  (6%)
- (e)  $y' - 7y = -2y^2$  (6%)

2. Solve the following equation by Laplace transform method.

$$y'' - 2y' + y = e^t, \quad y(0) = 1, \quad y'(0) = 0 \quad (14\%)$$

3. If  $A$  has eigenvalues 0 and 1, corresponding to the eigenvectors,

$$\begin{bmatrix} 1 \\ 2 \end{bmatrix} \text{ and } \begin{bmatrix} 2 \\ -1 \end{bmatrix}$$

respectively,

- (a) What is the matrix  $A$ ? (7%)
- (b) What will be the eigenvalues and eigenvectors of  $A^2$ ? (8%)

4. Let  $W$  be a subspace of a vector space  $V$ . Consider the following three subsets of  $W$ : (20%)

$$\begin{aligned} S_1 &= \{\mathbf{u}_1, \mathbf{u}_2, \dots, \mathbf{u}_7\} \\ S_2 &= \{\mathbf{v}_1, \mathbf{v}_2, \dots, \mathbf{v}_{10}\} \\ S_3 &= \{\mathbf{v}_1, \mathbf{v}_2, \dots, \mathbf{v}_{10}, \mathbf{v}_{11}, \mathbf{v}_{12}\} \end{aligned}$$

It is understood that each of  $S_1$ ,  $S_2$ , and  $S_3$  is a set of distinct vectors of  $V$ . Note that  $S_2$  is a subset of  $S_3$  but there are no other known relationships among the vectors in  $S_1$ ,  $S_2$ , and  $S_3$ . Here is a list of possible responses (*answers*) to the questions below ((i) – (v)):

- (A) ... is linearly independent.
- (B) ... is linearly dependent.
- (C) ... is not a basis for  $W$ .
- (D) ... spans  $W$ .
- (E) ... does not span  $W$ .
- (F) The hypothesis provides no information about the set mentioned in the conclusion.

Fill in the blanks (on your answer sheet) with a list of all those possible responses (*answers*) that can be justifiably concluded.

- (i) If  $S_1$  spans  $W$ , then  $S_2$  \_\_\_\_\_
- (ii) If  $S_2$  spans  $W$ , then  $S_3$  \_\_\_\_\_
- (iii) If  $S_3$  is a basis for  $W$ , then  $S_1$  \_\_\_\_\_
- (iv) If  $S_3$  is linearly independent, then  $S_2$  \_\_\_\_\_
- (v) If  $S_2$  is linearly independent, then  $S_3$  \_\_\_\_\_



5. Let

$$A = \begin{bmatrix} 1 & 2 & -3 \\ 2 & -1 & 4 \\ 4 & 3 & -2 \end{bmatrix}$$

Find a basis for

- (a) the column space of  $A$ . (5%)
- (b) the row space of  $A$ . (5%)
- (c) the null space of  $A$ . (5%)



注意：全部四題，總計一百分。

1 (25%) The switch in the circuit shown in Figure 1 has been in position 1 for a very long time. At  $t = 0$  it was placed in position 2 for 8 ms and then replaced into position 1. Find

- (a)  $dv_C/dt(0^+)$ , the initial rate of change of  $v_C$ .
- (b)  $v_C$  at  $t = 8$  ms.
- (c) The last time at which  $v_C$  is 10V.
- (d) The last time at which  $i = 20$  mA.
- (e) The time at which  $i = -2$  mA.

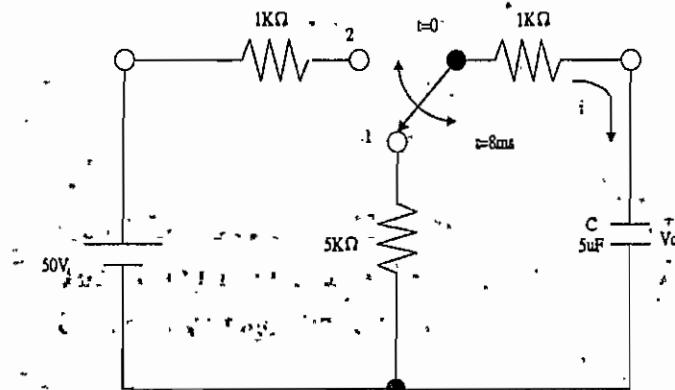


Figure 1

2.(25%) For the circuit shown in Figure 2, find

- (a) I.
- (b)  $V_D$ .
- (c) The phase angle between the  $Y_A$  and the  $Y_B$  waveforms.
- (d) The equivalent simple series circuit at this frequency.

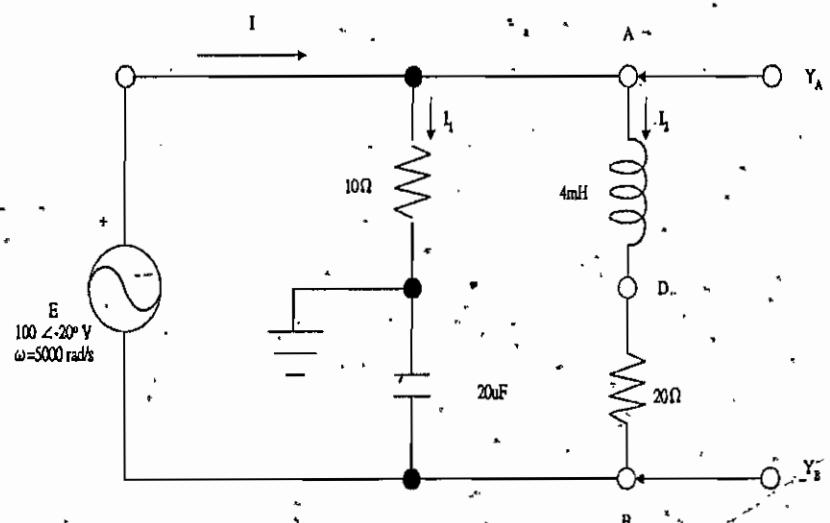


Figure 2



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八十七學年度研究所博士班入學考試試題

所別：工程技術研究所

科目：電路學

3. (25%) The network shown in Figure 3 is excited as follows:

$$i_1(t) = \cos(t) \quad i_2(t) = 2\cos(t + \pi/2)$$

Find the sinusoidal steady-state response for  $v_1(t)$  and  $v_2(t)$ .

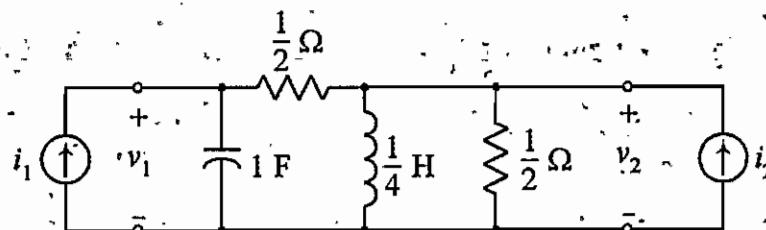


Figure 3

4. (25%) In the circuit of Figure 4, the initial voltage across the capacitor  $v(0) = 0$ .

The switch is closed at  $t = 0$ . Assume the diode is ideal. Find and sketch the capacitor voltage as a function of time.

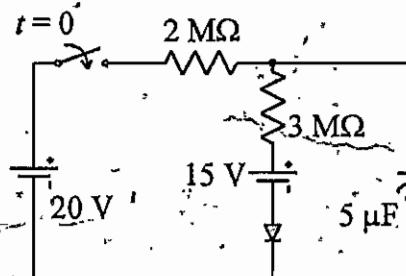


Figure 4



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所別：工程技術研究所

科目：自動控制

1. Briefly explain (or definition) the following terminologies (20%)

- (a) linear causal systems. (5%)
- (b) Routh-Hurwitz criterion. (5%)
- (c) Nyquist criterion. (5%)
- (d) overshoot percentage. (5%)

2. Find the fundamental matrix and state transition matrix of the following equation.

$$\begin{bmatrix} \dot{x}_1 \\ \dot{x}_2 \end{bmatrix} = \begin{bmatrix} -1 & e^{-t} \\ 0 & -1 \end{bmatrix} \begin{bmatrix} x_1 \\ x_2 \end{bmatrix} \quad (10\%)$$

3. Why does the root locus of  $1 + kG(s)H(s) = 0$ , begin ( $k = 0$ ) at the poles of  $G(s)H(s)$  and the end ( $k = \infty$ ) at the zeros of  $G(s)H(s)$ . (10%)

4. Briefly explain how to find the static error constant from the Bode magnitude plot. (10%)

5. Determine whether the following state equation is stable or not. (15%)

$$\begin{bmatrix} \dot{x}_1 \\ \dot{x}_2 \end{bmatrix} = \begin{bmatrix} -1 & 0 \\ e^{2t} & -3 \end{bmatrix} \begin{bmatrix} x_1 \\ x_2 \end{bmatrix}$$

6. The forward transfer function of a unity feedback system is given by

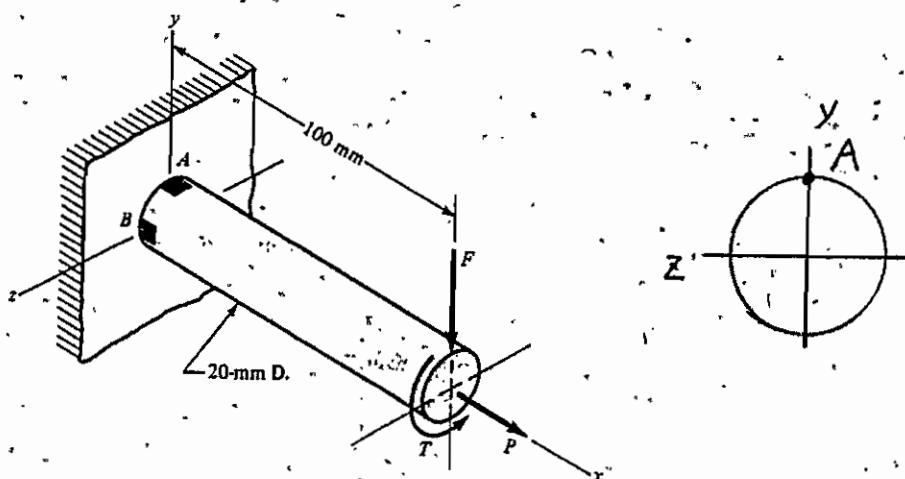
$$G(s) = \frac{50(s+\alpha)}{s(s+10)(s+30)}$$

Design the value of  $\alpha$  so that the closed-loop poles will have a damping ratio of 0.5. For this value of  $\alpha$ , obtain the response to a unit step. (20%)

7. Given the plant,  $G(s) = \frac{y(s)}{u(s)} = \frac{10}{(s+1)(s+2)}$ , design the state feedback for the plant represented in cascade form to yield a 15% overshoot with a settling time of 0.5 second: (15%)



1. The 20 mm diameter bar shown below is made of AISI 1006 cold drawn street with yielding stress 330 Mpa. The bar is loaded by the forces  $F=0.55 \text{ KN}$ ,  $p=8.0 \text{ KN}$ , and  $T=30 \text{ Nm}$ .
- (a) Determine the state of stress at point A. [15%]
- (b) Determine the factor of safety against yielding at point A by the maximum shear stress theory. [10%]



2. A component undergoes a cyclic stress with a maximum value of 110 ksi and a minimum value of 10 ksi. The component is made from a steel with an ultimate strength,  $S_u$ , of 150 ksi, an endurance limit,  $S_e$ , of 60 ksi, and a fully reversed stress at 1000 cycles,  $S_{1000}$ , of 110 ksi. Using the modified Goodman relationship, determine the life of the component.

[25%]



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所別：工程技術研究所

科目：機械設計

3. Find the friction power loss for a Petroff bearing 100 mm in diameter and 150 mm long. Radial clearance  $c$  is 0.05 mm. Speed is 900 rpm. SAE oil 10 is used at 175°F. [25%]

$$\mu = 0.000\ 001\ 03 \text{ lb sec/in.}^2$$

$$1 \frac{\text{lb sec}}{\text{in.}^2} = \frac{1}{145} \frac{\text{Nsec}}{\text{mm}^2}$$

$$\frac{F_1}{\mu U} \left( \frac{c}{r} \right) = 2\pi \quad (\text{Petroff's bearing Equation})$$

4. Two module 3 gears are to be mounted on a center distance of 384 mm. The speed ratio is 7:9. Find the number of teeth in each gear. [25%]

$$m_o = \frac{d}{N}$$



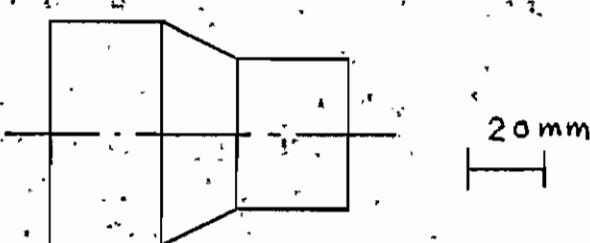
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八十七學年度研究所博士班入學考試試題

所別：工程技術研究所

科目：機械製造

1. 欲製造如圖所示之金屬工件若干個，試由製造成本、工件成品之機械性質（強度等）、尺寸精度、表面光度等方面，說明與比較以(1)車削；(2)鍛造；(3)焊接；(4)鑄造；(5)粉末冶金等五種製法之優缺點。(50%)



2. 試說明以何種製造程序可製作一長  $200\mu m$ ，內徑  $30\mu m$  外徑  $50\mu m$  之銅管，並比較各方法之優缺點。又如果管內壁之粗糙度希望在  $R_{\text{max}} 0.1\mu m$  以下，當如何處理？(25%)
3. 試說明雷射於機械製造工程之應用現況及未來發展。(25%)



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八十七學年度研究所博士班入學考試試題

所別：工程技術研究所

科目：材料科學概論

請依題序將答案寫在答案卷

20% 1. Plot and describe the point and line defects of the crystal.

20% 2. How to protect a metal to prevent it from the corrosion, please describe the optimum methods.

20% 3. Explain the following strengthening mechanisms

(a) precipitation hardening 4%

(b) strengthening by grain size reduction 4%

(c) solid-solution strengthening 4%

(d) strain hardening 4%

(e) texture hardening 4%

20% 4. Explain the following terms

(a) Glass transition temperature 4%

(b) Fick's second law 4%

(c) Transition band 4%

(d) Zener formula 4%

(e) Amorphous metal 4%

20% 5.(a) Explain the purpose and principle of zone melting. 10%

(b) Continuous cooling transformation (C-C-T) diagram is important for a heat treatment engineer. Explain how to make a C-C-T diagram of a new steel alloy in detail. 10%



1. (20 points)

Nitrogen gas is being withdrawn from a  $0.15 \text{ m}^3$  cylinder at the rate of  $10 \text{ mol/min}$ . The cylinder initially contains the gas at a pressure of  $100 \text{ bar}$  and  $170 \text{ K}$ . The cylinder is well insulated and there is a negligible heat transfer between the cylinder walls and the gas. What will be the temperature and pressure of the gas in the cylinder after  $50 \text{ minutes}$ ?

Assume that nitrogen is an ideal gas. Gas constant,  $R$ , is  $8.314 \times 10^{-5} \text{ bar m}^3/\text{mol K}$ .  $C_p (\text{J/mol K}) = 27.2 + 4.2 \times 10^{-3} T$ .

2. (30 points)

Find  $\Delta U$ ,  $\Delta H$ ,  $Q$ ,  $W$ , and  $\Delta S$  for the following processes, all performed on  $1 \text{ g mol}$  of an ideal gas with  $C_v = 5/2 R$ .

- A constant-pressure heating from  $25^\circ\text{C}$  to  $100^\circ\text{C}$
- An adiabatic reversible expansion from  $200^\circ\text{C}$ ,  $5 \text{ atm}$  to  $1 \text{ atm}$
- An isothermal reversible expansion from  $200^\circ\text{C}$ ,  $5 \text{ atm}$  to  $1 \text{ atm}$
- A constant-volume heating from  $25^\circ\text{C}$ ,  $1 \text{ atm}$  to  $100^\circ\text{C}$



3. (30 points)

In large steam power plants, the feedwater is frequently heated in closed feedwater heaters, which are basically heat exchangers, by steam extracted from the turbine at some stage. Steam enters the feedwater heater at 1 MPa and 200°C and leaves as saturated liquid at the same pressure. Feedwater enters the heater at 2.5 MPa and 50°C and leaves 10°C below the exit temperature of the steam. Neglecting any heat losses from the outer surfaces of the heater, determine (a) the ratio of the mass flow rates of the extracted steam and the feedwater and (b) the reversible work for this process per unit mass of the feedwater. Assume the surroundings to be at 25°C, and the properties of compressed liquid can be approximated using saturated liquid's data under the same temperature. (15 points each)

4. (20 points)

- (1) Fig. 1 is the schematic of a combined cycle plant. Under the operating condition shown in the figure, its overall efficiency is only 34.2% which means that 65.8% of the energy of the fuel is wasted. Please propose some feasible ideas about how to improve its overall efficiency. (10 points)
- (2) Is it possible to maintain a pressure of 10 kPa in a condenser which is being cooled by river water entering at 20°C? Why? (10 points)

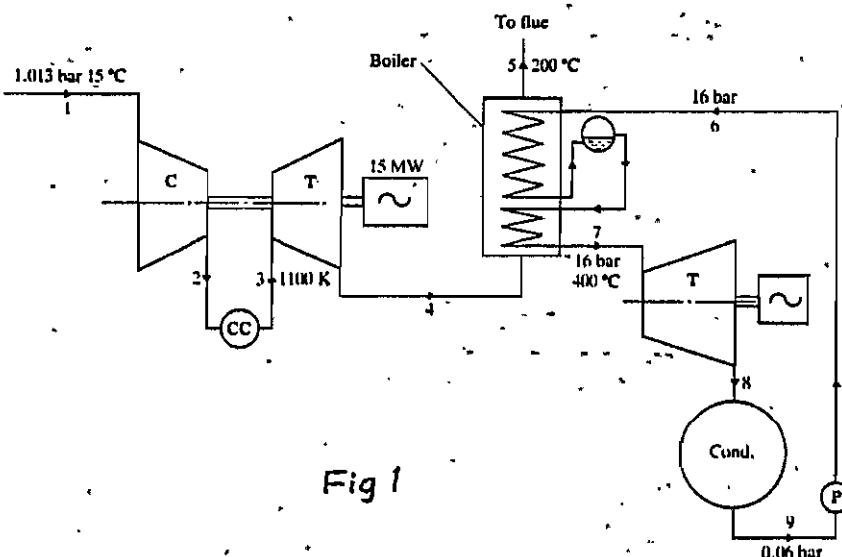


Fig 1



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八十七學年度研究所博士班入學考試試題

所別：工程技術研究所

科目：熱力學

Temp. °C T	Sat. press. kPa $P_{in}$	Specific volume m³/kg		Internal energy kJ/kg				Enthalpy kJ/kg			Entropy kJ/(kg · K)		
		Sat. liquid $v_f$	Sat. vapor $v_k$	Sat. liquid $u_f$	Evap. $u_{fR}$	Sat. vapor $u_k$	Sat. liquid $h_f$	Evap. $h_{fR}$	Sat. vapor $h_k$	Sat. liquid $s_f$	Evap. $s_{fR}$	Sat. vapor $s_k$	
0.01	0.6113	0.001000	206.14	0.0	2375.3	2375.3	0.01	2501.3	2501.4	0.000	9.1562	9.1562	
5	0.8721	0.001000	147.12	20.97	2361.3	2382.3	20.98	2489.6	2510.6	0.0761	8.9496	9.0257	
10	1.2276	0.001000	106.38	42.00	2347.2	2389.2	42.01	2477.7	2519.8	0.1510	8.7498	8.9008	
15	1.7051	0.001001	77.93	62.99	2333.1	2396.1	62.99	2465.9	2528.9	0.2245	8.5569	8.7814	
20	2.339	0.001002	57.79	83.95	2319.0	2402.9	83.96	2454.1	2538.1	0.2966	8.3706	8.6672	
25	3.169	0.001003	43.36	104.88	2304.9	2409.8	104.89	2442.3	2547.2	0.3674	8.1905	8.5580	
30	4.246	0.001004	32.89	125.78	2290.8	2416.6	125.79	2430.5	2556.3	0.4369	8.0164	8.4533	
35	5.628	0.001006	25.22	146.67	2276.7	2423.4	146.68	2418.6	2565.3	0.5053	7.8478	8.3531	
40	7.384	0.001008	19.52	167.56	2262.6	2430.1	167.57	2406.7	2574.3	0.5725	7.6845	8.2570	
45	9.593	0.001010	15.26	188.44	2248.4	2436.8	188.45	2394.8	2583.2	0.6387	7.5261	8.1648	
50	12.349	0.001012	12.03	209.32	2234.2	2443.5	209.33	2382.7	2592.1	0.7038	7.3725	8.0763	
55	15.758	0.001015	9.568	230.21	2219.9	2450.1	230.23	2370.7	2600.9	0.7679	7.2234	7.9913	
60	19.940	0.001017	7.671	251.11	2205.5	2456.6	251.13	2358.5	2609.6	0.8312	7.0784	7.9096	
Sat. press. MPa													
100	0.10135	0.001044	1.6729	418.94	2087.6	2506.5	419.04	2257.0	2676.1	1.3069	6.0480	7.3549	
105	0.12082	0.001048	1.4194	440.02	2072.3	2512.4	440.15	2243.7	2683.8	1.3630	5.9328	7.2958	
110	0.14327	0.001052	1.2102	461.14	2057.0	2518.1	461.30	2230.2	2691.5	1.4185	5.8202	7.2387	
115	0.16906	0.001056	1.0366	482.30	2041.4	2523.7	482.48	2216.5	2699.0	1.4734	5.7100	7.1833	
120	0.19853	0.001060	0.8919	503.50	2025.8	2529.3	503.71	2202.6	2706.3	1.5276	5.6020	7.1296	
125	0.2321	0.001065	0.7706	524.74	2009.9	2534.6	524.99	2188.5	2713.5	1.5813	5.4962	7.0775	
130	0.2701	0.001070	0.6685	546.02	1993.9	2539.9	546.31	2174.2	2720.5	1.6344	5.3925	7.0269	
135	0.3130	0.001075	0.5822	567.35	1977.7	2545.0	567.69	2159.6	2727.3	1.6870	5.2907	6.9777	
140	0.3613	0.001080	0.5089	588.74	1961.3	2550.0	589.13	2144.7	2733.9	1.7391	5.1908	6.9299	
145	0.4154	0.001085	0.4463	610.18	1944.7	2554.9	610.63	2129.6	2740.3	1.7907	5.0926	6.8833	
150	0.4758	0.001091	0.3928	631.68	1927.9	2559.5	632.20	2114.3	2746.5	1.8418	4.9960	6.8379	
155	0.5431	0.001096	0.3468	653.24	1910.8	2564.1	653.84	2098.6	2752.4	1.8925	4.9010	6.7935	
160	0.6178	0.001102	0.3071	674.87	1893.5	2568.4	675.55	2082.6	2758.1	1.9427	4.8075	6.7502	
165	0.7005	0.001108	0.2727	696.56	1876.0	2572.5	697.34	2066.2	2763.5	1.9925	4.7153	6.7078	
170	0.7917	0.001114	0.2428	718.33	1858.1	2576.5	719.21	2049.5	2768.7	2.0419	4.6244	6.6663	

Press. temp. MPa $P_{in}$	Sat. temp. °C $T_{in}$	Specific volume m³/kg		Internal energy kJ/kg				Enthalpy kJ/kg			Entropy kJ/(kg · K)		
		Sat. liquid $v_f$	Sat. vapor $v_k$	Sat. liquid $u_f$	Evap. $u_{fR}$	Sat. vapor $u_k$	Sat. liquid $h_f$	Evap. $h_{fR}$	Sat. vapor $h_k$	Sat. liquid $s_f$	Evap. $s_{fR}$	Sat. vapor $s_k$	
0.80	170.43	0.001115	0.2404	720.22	1856.6	2576.8	721.11	2048.0	2769.1	2.0462	4.6166	6.6628	
0.85	172.96	0.001118	0.2270	731.27	1847.4	2578.7	732.22	2039.4	2771.6	2.0710	4.5711	6.6421	
0.90	175.38	0.001121	0.2150	741.83	1838.6	2580.5	742.83	2031.1	2773.9	2.0946	4.5280	6.6226	
0.95	177.69	0.001124	0.2042	751.95	1830.2	2582.1	753.02	2023.1	2776.1	2.1172	4.4869	6.6041	
1.00	179.91	0.001127	0.19444	761.68	1822.0	2583.6	762.81	2015.3	2778.1	2.1387	4.4478	6.5865	
1.10	184.09	0.001133	0.17753	780.09	1806.3	2586.4	781.34	2000.4	2781.7	2.1792	4.3744	6.5536	
1.20	187.99	0.001139	0.16333	797.29	1791.5	2588.8	798.65	1986.2	2784.8	2.2166	4.3067	6.5233	

$t$ °C	$v$ m³/kg	$u$ kJ/kg	$h$ kJ/kg	$s$ kJ/(kg · K)
$P = 1.00 \text{ MPa (} 179.91^\circ\text{C)}$				
Sat.	0.19444	2583.6	2778.1	6.5865
200	0.2060	2621.9	2827.9	6.6940
250	0.2327	2709.9	2942.6	6.9247
300	0.2579	2793.2	3051.2	7.1229
350	0.2825	2875.2	3157.7	7.3011
400	0.3068	2957.3	3263.9	7.4651
500	0.3541	3124.4	3478.5	7.7622



下列各題，每題10分

1. The complete combustion of 1 mole ethyl alcohol(*l*) produces heat of 326.7 kcal. The heat of formation of CO<sub>2</sub> and water are 97.0 kcal and 68.4 kcal, respectively. Calculate the heat of formation of ethyl alcohol(*l*).
2. Define in detail (A)Pauli exclusion principle and (B)Ionization energy.
3. The production of NO by reaction of N<sub>2</sub> and O<sub>2</sub> in an automobile engine is an important source of nitrogen oxide pollution. At 1000°C the reaction  $N_{2(g)} + O_{2(g)} \rightleftharpoons 2NO_{(g)}$  has  $k_p = 4.8 \times 10^{-7}$ . Suppose that the partial pressure of N<sub>2</sub> and O<sub>2</sub> in the cylinder of an engine after gasoline vapor has been ignited are  $p_{N_2} = 33.6$  atm and  $p_{O_2} = 4.0$  atm. Assume the temperature is 1000°C. Calculate the partial pressure of NO in the mixture if the system has reached equilibrium.
4. What gives stability to hydrophobic colloids (particulates will not settle easily) in water? How to destabilize these colloids?
5. How many grams of HCl gas would have to be dissolved in 500ml of 1.0M NaCH<sub>3</sub>COO to give a solution having a pH = 4.74?  
 $K_a = 1.8 \times 10^{-5}$



6. A weather balloon is loosely inflated with helium at a pressure of 1.0atm(=760torr) and a temperature of 20°C. The gas volume is 2.2m<sup>3</sup>. At an elevation of 20,000 ft, the atmospheric pressure is down to 380 torr and the helium has expanded, being under no restraint from confining bag. At this elevation the gas temperature is -48°C. What is the gas volume now?
7. Two charged, parallel, flat conducting surfaces are spaced  $d = 1.00\text{cm}$  apart and produce a potential difference  $\Delta V=625\text{V}$  between them. An electron is projected from one surface directly toward the second. What is the initial speed of the electron if it comes to rest just at the second surface?
8. A solenoid having an inductance of  $6.30\ \mu\text{H}$  is connected in series with a  $1.20\ \text{k}\Omega$  resistor. (a) If a 14.0V battery is switched across the pair, how long will it take for the current through the resistor to reach 80.0% of its final value? (b) What is the current through the resistor at time  $t = 1.0\ \tau_L$ ?
9. The mean lifetime of stationary muons is measured to be  $2.2\ \mu\text{s}$ . The mean lifetime of high-speed muons in a burst of cosmic rays observed from Earth is measured to be  $16\ \mu\text{s}$ . Find the speed of these cosmic-ray muons relative to Earth.
10. Calculate the energy released in the fission reaction  

$$^{235}\text{U} + n \rightarrow ^{141}\text{Cs} + ^{93}\text{Rb} + 2n$$
- Needed atomic and particle masses are
- |                   |             |                  |            |
|-------------------|-------------|------------------|------------|
| $^{235}\text{U}$  | 235.04392 u | $^{93}\text{Rb}$ | 92.92157 u |
| $^{141}\text{Cs}$ | 140.91963 u | $n$              | 1.00867 u  |
- (Hint:  $u=932\text{MeV}$ )

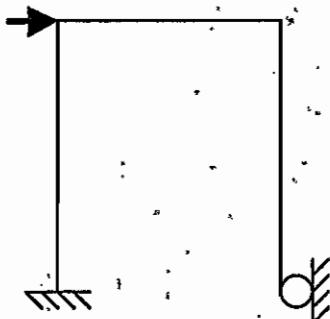


一、何謂莫氏圓(Mohr's circle)？其用途為何？證明之！[13分]

二、房屋結構可分為基礎、柱、梁及樓板，如果硬要討論各別的重要性的話，  
請排列其重要性順序並說明之。[12分]

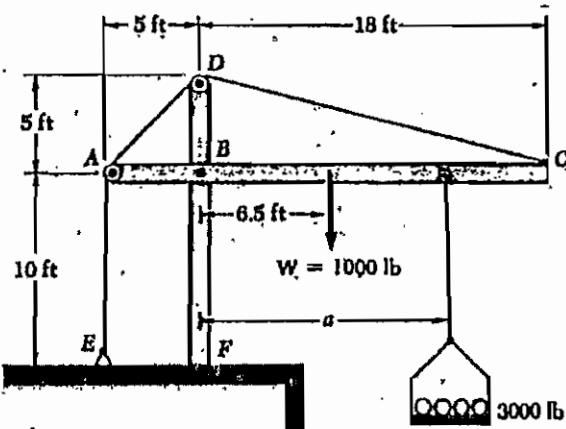
三、挑高的房子為何有安全顧慮？[13分]

四、以直覺繪出下圖所示結構之變形，請先繪原圖於答案卷上，再於該圖上繪出  
變形之圖形，以便與原圖對照。[12分]





五、如圖所示之吊架之水平桿件  $ABC$  重 1000 磅，且由一個鉸接  $B$  和一根纜繩  $EADC$  所支撐。由於纜繩跨過在  $A$  及  $D$  點之滑輪，纜繩所有部份之張力可假設為大小均相同。當以此吊架舉起一個 3000 磅重的重物（與垂直桿件  $DF$  之距離  $a = 12 \text{ ft}$ ）時，求（1）纜繩中之張力 [10 分]，（2）在  $B$  點的反力之水平與垂直分量 [15 分]。



六、一根 T 型斷面之懸臂梁，受一個大小為 10 kN 之傾斜力作用於自由端之上緣（如下圖），計算位於梁腹之  $A$  及  $B$  點處之主應力  $\sigma_1$  和  $\sigma_2$ ，以及最大之 in-plane 剪應力  $\tau_{max}$ 。[25 分]

